# Soret Effect on Free Convective Heat and Mass Transfer through a Porous Medium in a Vertical Channel 

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#### Abstract

we made an attempt to analyze the soret effect on convective heat and mass transfer through a porous medium in a vertical channel. The dissipative effects are also taken in to account . Assuming Eckkert number much less than 1,the governing equations are solved and the velocity, temperature, concentration, shear stress and rate of heat and mass transfer are analyzed for various parameters.


Key words: Heat and mass transfer, Porous medium, Soret effect.

## I. INTRODUCTION

Convective heat transfer in a porous medium has been the subject of intensive studies for the last few decades owing to its applications in different fields such as Chemical Engineering, Geothermal, Petroleum and Reservoir Engineering, Environmental protection ,thermal insulation cooling, Processing the food etc .The majority of the studies pertain to fluid flow and heat transfer in porous medium based on the Darcy flow model(5). Darcy's equation give satisfactory results for closely packed porous medium but does not explain the flow through sparsely distributed porous medium in later situation, Brinkman [3] proposed an alternate model by adding a term which accounts for the viscous shear in addition to the Darcy's equation.The first theoretical investigation of natural convection in a porous enclosure using Brinkman model was made by Chan et al [4], Vafai and Tien [19], Vafai [18], Kim and Vafai [10]. They have worked on the problem of convective heat transfer in porous media relaxing some or all the limitations of Darcy's model . Later on a series of investigations were carried out using the Brinkman model by a few authors notably, Poulikakos and Bejan[13],Tong and Subramanian[17], Prasad and Tuntomo[14]. Forcheiner [7] extended Darcy model to the natural convection flow through porous medium.

Convection flows driven by temperature and concentration differences have been studied extensively in the past and various extensions of the problems have been reported in the literature[1],[9],[11],[15],[16],[20],[21] with both concentration and temperature interacting simultaneously, the convection can become quite complex .

Since many industrially and environmentally relevant fluids are not pure, it has been suggested that more attention should be paid to convective phenomena which can occur in mixtures, but are not present in common fluids such as air or water. Applications involving liquid mixtures include the casting of alloys, ground water pollutant migration and separation operations. In all of these situations multi-component liquids can undergo natural convection driven by buoyancy force resulting from simultaneous temperature and species gradients. In the case of binary mixtures, species gradients can be established by the applied solute boundary conditions such as species rejection associated with alloys casting or can be induced by coupled transport mechanisms such as Soret (thermo) diffusion. In the case of Soret diffusion, species gradients are established in an otherwise uniform concentration mixture in accordance with the onsager reciprocal relationships. Recently some importance has been attached to the Benard problem in a two component system in which an initially homogeneous mixture is subjected to a temperature gradient. Then thermal diffusion known as the Soret effect takes place and as a result a mass fraction distribution is established in the liquid layer. The sense of migration of the molecular species be determined by the sign of Soret coefficient. Keeping this view several authors have investigated the Soret effect under varied conditions [2],[6],[8],[12].

## II. FORMULATION OF THE PROBLEM

We consider the flow of a viscous, incompressible fluid through a porous medium confined in a vertical channel bounded by two flat plates. We choose a rectangular cartesian coordinate system $\mathrm{O}(\mathrm{x}, \mathrm{y}, \mathrm{z})$ with the plates in the $x$-y plane. The $z$-axis is taken normal to the plane of the plates. The walls are maintained at constant temperature $T_{1} \& T_{2}$ and constant concentrations $C_{1} \& C_{2}$ in the presence of constant heat sources of
strength Q. All the field properties are assumed constant except that the influence of the density variation with the temperature and concentration is considered only in the body force term. The viscous dissipation and Darcy dissipation are taken into account. Also we take into consideration the thermal diffusion(Soret effect). A linear density variation is assumed with $\rho_{0}, T_{0} \& C_{0}$ being the density, temperature and concentration of the fluid in the equilibrium state. The equations governing flow, heat and mass transfer in Cartesian coordinate $\mathrm{O}(\mathrm{x}, \mathrm{y}, \mathrm{z})$ in the non-dimensional form are

$$
\begin{align*}
& \frac{\partial^{2} u}{\partial y^{2}}-D^{-1} u+G(\theta+N c)=0  \tag{1}\\
& \frac{\partial^{2} \theta}{\partial y^{2}}+E c \cdot \operatorname{Pr}\left\lceil\left(\frac{\partial u}{\partial y}\right)^{2}\right\rceil+E c \cdot \operatorname{Pr} D^{-1} u^{2}+\alpha=0  \tag{2}\\
& \frac{\partial^{2} C}{\partial y^{2}}+\left(\frac{S_{c} S_{0}}{N}\right) \frac{\partial^{2} \theta}{\partial y^{2}}=0 \tag{3}
\end{align*}
$$

where $G=\frac{\beta g\left(T_{1}-T_{0}\right) h^{3}}{v^{2}}($ Grashof number $) \quad \mathrm{D}^{-1}=\frac{h^{2}}{k}($ Darcy number $)$

$$
\begin{array}{cc}
\mathrm{P}=\frac{\mu C_{p}}{k_{1}}(\text { Prandtl number }) & \mathrm{E}_{\mathrm{c}}=\frac{v^{2}}{h^{2}\left(T_{1}-T_{0}\right) C_{p}} \text { (Eckert number). } \\
\alpha=\frac{Q h^{2}}{k_{1}} \text { (heat source parameter) } & S c=\frac{v}{D_{1}}(\text { Schmidt number }) \\
S_{0}=\frac{\beta^{\cdot} k_{11}}{\beta v}(\text { Soret parameter }) &
\end{array}
$$

The corresponding boundary conditions are

$$
\begin{align*}
& u=0 \text { on } y= \pm 1 \\
& \theta=1, C=1 \text { on } y=-1  \tag{4}\\
& \theta=-1, C=-1 \text { on } y=1
\end{align*}
$$

Assuming $\operatorname{Ec}(\ll 1)$ to be small, we take the asymptotic expansions of velocity, temperature and concentration as

$$
\begin{align*}
& u=u_{0}+E c u_{1}+O\left(E c^{2}\right) \\
& \theta=\theta_{0}+E c \theta_{1}+O\left(E c^{2}\right)  \tag{5}\\
& C=C_{0}+E c C_{1}+O\left(E c^{2}\right)
\end{align*}
$$

Substituting (5) in (1)-(3), the zeroth order equations are

$$
\begin{align*}
& \frac{\partial^{2} u_{0}}{\partial y^{2}}-D^{-1} u_{0}=-G\left(\theta_{0}+N C_{0}\right)  \tag{6}\\
& \frac{\partial^{2} \theta_{0}}{\partial y^{2}}+\alpha=0  \tag{7}\\
& \frac{\partial^{2} C_{0}}{\partial y^{2}}+\left(S c S_{o} / N\right) \frac{\partial^{2} \theta_{0}}{\partial y^{2}}=0 \tag{8}
\end{align*}
$$

The respective boundary conditions are

$$
\begin{align*}
& u_{0}=0 \quad \text { on } \quad y= \pm 1 \\
& \theta_{0}=1, C_{0}=1 \quad \text { on } \quad y=-1 \\
& \theta_{0}=-1, C_{0}=-1 \quad \text { on } \quad y=1 \tag{9}
\end{align*}
$$

The equations to the first order are

$$
\begin{align*}
& \frac{\partial^{2} u_{1}}{\partial y^{2}}-D^{-1} u_{1}=-G\left(\theta_{1}+N C_{1}\right)  \tag{10}\\
& \frac{\partial^{2} \theta_{1}}{\partial y^{2}}+P\left(\frac{\partial u_{0}}{\partial y}\right)^{2}+P D^{-1} u_{0}^{2}=0  \tag{11}\\
& \frac{\partial^{2} C_{1}}{\partial y^{2}}+(S c S o / N) \frac{\partial^{2} \theta_{1}}{\partial y^{2}}=0 \tag{12}
\end{align*}
$$

with the boundary conditions

$$
\begin{equation*}
u_{1}=0, \theta_{1}=0, C_{1}=0 \quad \text { on } \quad y= \pm 1 \tag{13}
\end{equation*}
$$

Solving the differential equations (6)-(8) and (10)-(12) with respective boundary conditions (9) and (13) we obtain

$$
\begin{aligned}
& u_{o}=a_{8} e^{h_{1} y}+a_{9} e^{-h_{1} y}+a_{5} y^{2}+a_{6} y+a_{7} \\
& \theta_{0}=\frac{\alpha}{2}\left(1-y^{2}\right)-h \\
& C_{0}=a_{1}\left(y^{2}-1\right)-y \\
& u_{1}=a_{62} e^{h_{1} y}+a_{63} e^{-h_{1} y}+a_{42} C h\left(2 h_{1} y\right)+a_{43} e^{-2 h_{1} y}+\left(a_{44} y^{2}+a_{45} y+a_{46}\right) \operatorname{Sh}\left(h_{1} y\right) \\
& +\left(a_{47} y^{3}+a_{48} y^{2}+a_{4^{4} 9} y+a_{50}\right) C h\left(h_{1} y\right)+\left(a_{51} y^{3}+a_{52} y^{2}+a_{53} y+a_{54}\right) e^{-h_{1} y} \\
& +a_{55} y^{6}+a_{56} y^{5}+a_{57} y^{4}+a_{58} y^{3}+a_{59} y^{2}+a_{60} y+a_{61} \\
& \theta_{1}=a_{10} C h\left(2 h_{1} y\right)+a_{11} e^{-2 h_{1} y}+\left(a_{12} y+a_{13}\right) \operatorname{Ch}\left(h_{1} y\right)+\left(a_{14} y^{2}+a_{15} y+a_{16}\right) \operatorname{Sh}\left(h_{1} y\right) \\
& +\left(a_{17} y^{2}+a_{18} y+a_{19}\right) e^{-h_{1} y}+a_{20} y^{4}+a_{21} y^{3}+a_{22} y^{2}+a_{23} y+a_{24} \\
& C_{1}=-\frac{S c S_{o}}{N}\left(a_{10} C h\left(2 h_{1} y\right)+a_{11} e^{-2 h_{1} y}+\left(a_{12} y+a_{13}\right) C h\left(h_{1} y\right)+\left(a_{14} y^{2}+a_{15} y\right.\right. \\
& \left.+a_{16}\right) S h\left(h_{1} y\right)+\left(a_{17} y^{2}+a_{18} y+a_{19}\right) e^{-h_{1} y}+a_{20} y^{4}+a_{21} y^{3}+a_{22} y^{2}+a_{23} y+a_{24}
\end{aligned}
$$

Where the constants $a_{1}, a_{2}, \ldots \ldots, a_{63}$ are constants.
The shear stress on the plates are given by

$$
\tau=\mu\left(\frac{\partial u}{\partial y}\right)_{y= \pm h}
$$

which in the non-dimensional form reduces to

$$
\begin{aligned}
&(\tau)_{y=-1}=u_{o}^{1}(-1)+E c u_{1}^{1}(-1) \\
&=b_{1}+E c b_{3} \\
&(\tau)_{y=1}= u_{o}^{1}(1)+E c u_{1}^{1}(1) \\
&= b_{2}+E c b_{4} \\
&(\tau)_{y=-1}=u_{o}^{1}(-1)+E c u_{1}^{1}(-1) \\
&=b_{1}+E c b_{3} \\
& \text { and }
\end{aligned}
$$

the corresponding expressions are

$$
\begin{aligned}
(\tau)_{y=1} & =u_{o}^{1}(1)+E c u_{1}^{1}(1) \\
& =b_{2}+E c b_{4}
\end{aligned}
$$

where the constants $b_{1}, b_{2}, b_{3}, b_{4}$ are given in the Appendix.
From the temperature field, the rate of heat transfer coefficient in terms of Nusselt Number( Nu ) is given by

$$
\begin{aligned}
N u & =\frac{q_{w} h}{K\left(T_{1}-T_{o}\right)}=\left(\frac{\partial \theta}{\partial y}\right)_{y= \pm h} \\
& =\left(\theta_{o}^{1}+E c \theta_{1}^{1}\right)_{y= \pm 1}
\end{aligned}
$$

And the corresponding expressions are

$$
\begin{aligned}
& (N u)_{y=-1}=b_{5}+E c b_{7} \\
& (N u)_{y=1}=b_{6}+E c b_{8}
\end{aligned}
$$

where the constants $\mathrm{b}_{5}, \mathrm{~b}_{6}, \mathrm{~b}_{7}, \mathrm{~b}_{8}$ are constants.
The rate of mass transfer (Sherwood number) on the plates are given by

$$
\begin{aligned}
S h & =\left(\frac{\partial C}{\partial y}\right)_{y= \pm 1} \\
& =\left(C_{0}^{1}+E c C_{1}^{1}\right)_{y= \pm 1}
\end{aligned}
$$

and the corresponding expressions are

$$
\begin{aligned}
& (S h)_{y=-1}=b_{9}+E c b_{11} \\
& \text { and } \\
& (S h)_{y=1}=b_{10}+E c b_{12}
\end{aligned}
$$

where the constants $\mathrm{b}_{9}, \mathrm{~b}_{10}, \mathrm{~b}_{11}$ and $\mathrm{b}_{12}$ are constants.

## III. MATHEMATICAL MODEL AND VALIDATION

The set of partial differential equations (6)-(8) \& (10)-(12) together with the boundary conditions (9) \& (13) are numerically solved by employing perturbation technique. The following section, the results are discussed in detail with the aid of plotted graphs and tables. We made an attempt to discuss the flow of a viscous incompressible fluid, heat and mass transfer of the flow through a porous medium. In our numerical simulation the default values of the parameters are $\mathrm{G}=2 \times 10^{3}, \mathrm{~N}=1, \mathrm{D}^{-1}=10^{3}, \alpha=2, \mathrm{Sc}=1.3, \mathrm{So}=$. In order to analyze the effects of various parameters on velocity, temperature and concentration profiles, several graphs are plotted.

The profiles for velocity (u) for variation in the different governing parameters $D^{-1}, N, \alpha, S c \&$ $S_{0}$ are presented in figs.(1)-(4). For all variations we find that the velocity gradually rises from its prescribed value at $y=-1$ to attain a maximum at $y=-0.6$ and then falls to its prescribed value at $y=1$. Lesser the permeability of the medium smaller the velocity in fluid region (fig.1). The effect of the molecular diffusivity on the velocity $u$ shows that when $\mathrm{Sc} \sim 0.24$ we find an enhancement in $u$. For higher Sc $\sim 0.6$ the velocity $u$ fluctuates with maximum at $y=-0.6$ (fig.2). When the Soret parameter So increases through positive values the velocity is in upward direction and enhance with So. When So increases through negative values $u$ changes from upwards to downwards as we move from the left boundary $y=-1$ to the right boundary $y=1$ and the maximum $u$ is attained at $\mathrm{y}=0.4$ (fig. 3 ). In the presence of a heat source the velocity is upwards and is downwards in the case of a heat sink, We find that the magnitude of $u$ enhances with increase in the intensity of either a heat source or a sink .In the case of a heat source maximum of $u$ is attained at $y=-0.6$ while in the case of heat sink the maximum occurs at $\mathrm{y}=0.6$ (fig.4).

The temperature distribution ( $\theta$ ) for different variations of the parameters is exhibited in figs.(5)(9). The temperature is positive or negative according as the actual temperature is greater or smaller than the mean temperature. We find that for variation in $\mathrm{D}^{-1}$, So, $\alpha>0, \mathrm{~N} \& S c$ we find that in the region $-1 \leq y \leq 0.5$ the actual temperature is greater than the mean temperature while in the region $0.5<y \leq 1$ the actual temperature is less than the mean temperature. The variation of $\theta$ with Darcy parameter $\mathrm{D}^{-1}$ shows that the temperature decreases in magnitude with increase in $\mathrm{D}^{-1} \leq 5 \times 10^{3}$ but for higher $\mathrm{D}^{-1} \geq 10^{4},|\theta|$ enhances in the fluid region(fig.5).When the Soret parameter So increases through positive values we find a reduction in $\theta$ but for negative values of So we notice an enhancement in $|\theta|($ fig. 6$)$.Fig. 7 shows the variation of $\theta$ with heat source parameter $\alpha$. With an increase in the intensity of heat source we find an enhancement in $|\theta|$ with maximum attained at $y=-0.2$. In the case of heat sink the actual temperature is greater than the mean temperature in the region $-0.8 \leq y \leq-0.5$ while in the remaining fluid region the actual temperature is less than the mean temperature. $|\theta|$ enhances with the intensity of the heat sink. Maximum $\theta$ is attained at $y=0.2$. When the concentration buoyancy force dominates over the thermal buoyancy force the magnitude of the temperature decreases irrespective of the directions of the buoyancy forces (fig.9). Also we find that an increase in the molecular diffusivity decreases the temperature in the fluid region (fig.8).

The concentration distribution (C) is shown in figs.(10)-(12). As in the case of temperature the concentration in the fluid region is positive or negative according as the actual concentration is greater or lesser than the mean concentration. The magnitude of the concentration enhances with $\mathrm{D}^{-1} \leq 5 \times 10^{3}$ but for higher $\mathrm{D}^{-}$ ${ }^{1} \geq 10^{4}$ we find a depreciation in $|\mathrm{C}|$ with maximum attained at $\mathrm{y}=-0.8$ (fig.10). When the Soret parameter So increases through positive values we find an enhancement in $|\mathrm{C}|$ with maximum at $\mathrm{y}=-0.4$ while for So increasing through negative values the actual concentration is less than the mean concentration with maximum at $\mathrm{y}=0$.(fig.11). The variation of C with $\alpha$ shows that C is positive for $\alpha>0$ and negative for $\alpha<0$. The magnitude of the concentration enhances with increase in the intensity of a heat source or a $\operatorname{sink}(f i g .12)$.

The shear stress ( $\tau$ ) on the boundaries is evaluated for different variations of $\mathrm{G}, \mathrm{D}^{-1}, \mathrm{~N}, \alpha, \mathrm{Sc} \&$ So and are presented in tables.1-4.We find that the shear stress on the boundaries are almost negative except for $\alpha<0$.It is noticed that the stress on both the plates increase with increase in thermal buoyancy G.A decrease in the permeability of the porous medium reduces the magnitude of the shear stress on both the plates .At $\mathrm{y}=1$ an increase in the intensity of the heat source reduces $|\tau|$ and increases it in the case of a heat sink .On the boundary $y=-1$ for $D^{-1} \leq 5 \times 10^{3}$ the magnitude of the shear stress increases with $\alpha \leq 3$ but for higher $\alpha \geq 4,|\tau|$ enhances. For $D^{-1} \geq 10^{4},|\tau|$ increases with all values of $\alpha$ while in the case of heat sink, $|\tau|$ increases with $|\alpha|(\alpha<0)$. When the concentration buoyancy dominates over the thermal buoyancy $|\tau|$ on both the boundaries increase when the forces act in the same direction, but when they act in opposing directions $|\tau|$ decreases with $|\mathrm{N}|$ at the left boundary and increases it at the right boundary. As the Soret parameter So increases through negative values $|\tau|$ enhances on both the boundaries. When So increases through positive values, $|\tau|$ enhances with So for $\mathrm{D}^{-1} \leq 5 \times 10^{3}$
and for higher $\mathrm{D}^{-1} \geq 10^{4}$, it increases with So at $\mathrm{y}=1$ while $|\tau|$ on $\mathrm{y}=-1$ decreases with $|\mathrm{So}|$ for all $\mathrm{D}^{-1}$.Also an increase in the molecular diffusivity enhances $|\tau|$ on both the plates(tables.1\&4).

The rate of heat transfer (Nusselt number) on the plates for different variations of the governing parameters are presented in tables.5-8. We find that on the plate $y=1$ for $\mathrm{D}^{-1} \leq 10^{3}$, the rate of heat transfer decreases with thermal buoyancy parameter $\mathrm{G} \leq 3 \times 10^{3}$ but for higher $\mathrm{G} \geq 5 \times 10^{3},|\mathrm{Nu}|$ increases. For $\mathrm{D}^{-}$ ${ }^{1} \geq 3 \times 10^{3}$ the rate of heat transfer increases with $G$ while at $y=-1$ for $D^{-1} \leq 5 \times 10^{3}$ it reduces with $G$ but for $D^{-}$ ${ }^{1} \geq 10^{4}$ it enhances with G.A decrease in the permeability of the porous medium reduces Nu on $\mathrm{y}=1$ while on $\mathrm{y}=-$ $1, \mathrm{Nu}$ decreases with $\mathrm{D}^{-1} \leq 3 \times 10^{3}$ and increases with $\mathrm{D}^{-1} \geq 5 \times 10^{3}$. When the two buoyancy forces are in the same direction the rate of heat transfer on both the boundaries enhance with increase N . But when they are in opposing directions the magnitude of Nu reduces with $|\mathrm{N}| \mathrm{On} \mathrm{y}=1$ and on $\mathrm{y}=-1,|\mathrm{Nu}|$ reduces with N for $\mathrm{D}^{-}$ ${ }^{1} \leq 10^{3}$ while for $\mathrm{D}^{-} 1 \geq 3 \times 10^{3}$ it increases with N . When the Soret parameter So increases through positive values the rate of heat transfer reduces with So while its magnitude enhances with So when it increases through negative values. An increase in the intensity of the heat source $\alpha \leq 5$, we find a depreciation in $|\mathrm{Nu}|$ for $\mathrm{D}^{-}$ ${ }^{1} \leq 3 \times 10^{3}$, while for $\mathrm{D}^{-1} 1 \geq 5 \times 10^{3},|\mathrm{Nu}|$ increases with $\alpha \geq 10$.In the case of heat sink we find that the magnitude of Nu enhances with $|\alpha|$.Also an increase in the molecular diffusivity reduces $|\mathrm{Nu}|$ on the boundary $\mathrm{y}=1$ and enhances it on the boundary $\mathrm{y}=-1$ (tables.5-8).

The rate of mass flux(Sherwood number) on the boundaries are presented in tables. 9-12 for different variations in the governing parameters. We find that for $\mathrm{D}^{-1} \leq 3 \times 10^{3}$ the rate of mass transfer reduces with $\mathrm{G} \leq 3 \times 10^{3}$ and enhances with $\mathrm{G} \geq 5 \times 10^{3}$ while for $\mathrm{D}^{-1} 1 \geq 5 \times 10^{3}$ we find a reduction in the rate of mass flux on $y=1$ and an enhancement in it on $y=-1$. A decrease in the permeability of the porous medium reduces Sh on both the boundaries while $|\mathrm{Sh}|$ decreases with $\mathrm{D}^{-1} \leq 3 \times 10^{3}$ and increases with
$\mathrm{D}^{-1} \geq 10^{3}$. When the buoyancy forces act in the same directions the rate of mass transfer reduces with N at $\mathrm{y}=1$ and enhances at $\mathrm{y}=-1$ but when they are in opposing directions Sh reduces with N . An increase in the intensity of the heat source increases Sh at $y=1$ and at $y=-1$ it decreases with $\alpha \leq 5$ and increases with $\alpha \geq 10$.A reversed effect is observed in the case of heat sink .The variation of Sh with So shows that an increase in So enhances Sh .Also an increase in the molecular diffusivity reduces the magnitude of Sh at $\mathrm{y}=1$ and increases it on $\mathrm{y}=-1$.


Fig. 1 u with $\mathrm{D}^{-1}$
$\mathrm{G}=10^{3}, \mathrm{~N}=1, \mathrm{Sc}=1.3, \mathrm{So}=0.5, \alpha=2$
$\begin{array}{ccccc} & \text { I } & \text { II } & \text { III } & \text { IV } \\ \text { D-1 } & 3 \times 10^{3} & 5 \times 10^{3} & 10^{4} & 2 \times 10^{4}\end{array}$


Fig. 2 Variation of u with Sc
$\mathrm{D}^{-1}=10^{3}, \mathrm{G}=10^{3}, \mathrm{~N}=1, \mathrm{So}=0.5$
$\begin{array}{lllll}\text { Sc } & 1.3 & 2.01 & 0.24 & 0.6\end{array}$


Fig. 3 u with So
$\mathrm{G}=10^{3}, \mathrm{D}^{-1}=10^{3}, \mathrm{Sc}=1.3, \mathrm{~N}=1, \alpha=2$

|  | I | II | III | IV |
| :---: | :---: | ---: | :---: | :---: |
| So | 0.5 | 1 | -0.5 | -1.0 |



Fig4 $u$ with $\alpha$
$\mathrm{G}=10^{3}, \mathrm{D}^{-1}=10^{3}, \mathrm{Sc}=1.3, \mathrm{~N}=1, \alpha=2$

|  | I | II | III | IV | V | VI |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\alpha$ | 2 | 5 | 10 | -2 | -5 | -10 |



Fig. $5 \quad \theta$ with $\mathrm{D}^{-1}$

$$
\begin{array}{cccccc}
\mathrm{G}=10^{3}, \mathrm{~N}=1, \mathrm{Sc}=1.3, & \mathrm{So}=0.5, \alpha=2 \\
& \text { I } & \text { II } & \text { III } & \text { IV } \\
& \mathrm{D}^{-1} & 3 \times 10^{3} & 5 \times 10^{3} & 10^{4} & 2 \times 10^{4}
\end{array}
$$


$\begin{array}{ccccrr} & & & & & \\ & & \text { Fig. } 6 & \theta & \text { with } & \text { So } \\ & \text { I } & \text { II } & \text { III } & \text { IV } \\ \text { So } & 0.5 & 1.0 & -0.5 & -1.0\end{array}$

 \begin{tabular}{llllllll}
\& Fig. 7 \& \multicolumn{2}{c}{$\theta$} \& with \& \multicolumn{2}{c}{$\alpha$} <br>
\& I \& II \& III \& IV \& V \& VI <br>
$\alpha$ \& 2 \& 5 \& 10 \& -2 \& -5 \& -10

 

<br>
\hline \& - \& 5 \& 10 \& -2 \& -5 \& -10
\end{tabular}

Fig. $9 \quad \theta$ with N
$\begin{array}{ccccr} & \text { I } & \text { II } & \text { III } & \text { IV } \\ \mathrm{N} & 1 & -0.5 & 2 & -0.8\end{array}$


Fig. $8 \quad \theta$ with Sc $\begin{array}{lllll}\text { Sc } & 1.3 & 2.01 & 0.24 & 0.6\end{array}$


Fig. $10 \quad \mathrm{C}$ with $\mathrm{D}^{-1}$
$\mathrm{G}=10^{3}, \mathrm{~N}=1, \mathrm{Sc}=1.3, \mathrm{So}=1.5$
$\mathrm{D}^{-1} 3 \times 10^{3} 5 \times 10^{3} 10^{4} 2 \times 10^{4}$


Fig. 11 C with So
$\mathrm{G}=10^{3}, \mathrm{D}^{-1}=10^{4}, \mathrm{~N}=1, \mathrm{Sc}=1.3$
I II III IV
$\begin{array}{lllll}\text { Sc } & 0.5 & 1.0 & -0.5 & -1.0\end{array}$


Fig. 12 C with $\alpha$
I II III IV V VI
$\begin{array}{lllllll}\alpha & 2 & 5 & 10 & -2 & -5 & -10\end{array}$

Table. 1
Shear stress ( $\tau$ ) at $\mathrm{y}=1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | 0.5399 | 2.5674 | 15.898 | 2.3661 | .5038 | 0.1183 | 0.6012 | -1.436 | -.5604 |
| $3 \times 10^{3}$ | -.0549 | -.2254 | -2.135 | -.4437 | .0612 | .0007 | -.0087 | .2839 | .1282 |
| $5 \times 10^{3}$ | -.0201 | -.0372 | -.2455 | -.0978 | .0419 | .0023 | -.0065 | .1012 | .0451 |
| $10^{4}$ | -.0199 | -.0307 | -.0315 | -.0236 | .0323 | .0006 | -.0045 | .0225 | .0089 |
| $2 \times 10^{4}$ | -.0253 | -.0496 | -.0871 | -.0197 | .0284 | -.0015 | -.0021 | -.0176 | -.0252 |

Table. 2
Shear Stress ( $\tau$ ) at $\mathrm{y}=-1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | -13.713 | -15.927 | -24.354 | -15.544 | 1.1961 | -1.431 | -5.038 | -17.981 | -16.625 |
| $3 \times 10^{3}$ | -6.4813 | -9.805 | -16.954 | -9.346 | 0.0377 | -0.937 | -1.979 | -8.776 | -8.385 |
| $5 \times 10^{3}$ | -2.4813 | -7.008 | -9.301 | -5.464 | -0.1851 | -0.574 | 0.4114 | -4.977 | -4.341 |
| $10^{4}$ | 2.4404 | -4.199 | -2.048 | -3.0327 | -0.1411 | -0.299 | 2.0041 | -3.165 | -2.335 |
| $2 \times 10^{4}$ | 1.5315 | 1.4182 | 5.2084 | -0.3618 | 0.1308 | 0.1292 | 3.6895 | -1.4708 | -0.472 |


| G | $2 \times 10^{3}$ | $3 \times 10^{3}$ | $5 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| N | 1 | 1 | 1 | 2 | -.5 | -.8 | 1 | 1 | 1 |
| Sc | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 2.01 | 0.24 | 0.6 |

Table. 3
Shear Stress ( $\tau$ ) at $\mathrm{y}=1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | -3.582 | 0.0565 | 8.1739 | 0.5399 | -.0782 | 0.7271 | -8.137 | -9.819 | -7.348 |
| $3 \times 10^{3}$ | 0.6854 | 0.0298 | -1.588 | -.0594 | -.0151 | .1422 | -9.238 | 11.181 | 14.756 |
| $5 \times 10^{3}$ | 0.2442 | 0.0340 | -0.562 | -.0202 | -.0159 | .1599 | 5.1483 | 9.656 | 11.238 |
| $10^{4}$ | 0.0849 | 0.0404 | -.2015 | -.0201 | -.0134 | .1484 | 2.3181 | 5.838 | 8.327 |
| $2 \times 10^{4}$ | -.201 | 0.0497 | -.0631 | -.0253 | -.0115 | .1299 | 1.2359 | 3.092 | 5.531 |

Table. 4
Shear Stress $(\tau)$ at $\mathrm{y}=-1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | -8.378 | 6.093 | 11.338 | - <br> 13.713 | -17.744 | -11.558 | 68.123 | 98.569 | 112.145 |
| $3 \times 10^{3}$ | -7.453 | 3.414 | -.0285 | -6.253 | -5.734 | 1.672 | 52.321 | 82.468 | 92.126 |
| $5 \times 10^{3}$ | -5.957 | 3.731 | -5.698 | -2.482 | 0.531 | 6.652 | 30.608 | 49.579 | 72.831 |
| $10^{4}$ | -6.021 | 4.499 | -8.251 | 2.441 | 2.483 | 8.533 | 20.338 | 33.591 | 50.067 |
| $2 \times 10^{4}$ | -6.701 | 5.619 | -9.515 | 1.532 | 4.318 | 9.212 | 11.696 | 19.554 | 29.543 |


| $\mathrm{S}_{0}$ | -.05 | 1 | -1 | 0.5 | 0.5 | 0.5 | 0.5 | 0.5 | 0.5 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $\alpha$ | 2 | 2 | 2 | 2 | 5 | 10 | -2 | -5 | -10 |

Table. 5
Nusselt Number (Nu) at $\mathrm{y}=1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | -32.684 | -3.029 | -52.44 | -35.02 | -89.739 | -65.114 | -3.419 | -79.679 | -45.958 |
| $3 \times 10^{3}$ | -4.058 | -5.517 | -10.49 | -4.855 | -3.601 | -3.239 | -3.488 | -5.068 | -4.704 |
| $5 \times 10^{3}$ | -3.273 | -3.611 | -4.699 | -3.532 | -3.096 | -3.026 | -3.054 | -3.6154 | -3.4967 |
| $10^{4}$ | -3.165 | -3.366 | -4.007 | -3.343 | -3.056 | -2.998 | -3.015 | -3.393 | -3.315 |

Table. 6
Nusselt Number (Nu) at $\mathrm{y}=-1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | 19.5503 | 1.0634 | -13.29 | 37.1287 | 38.3636 | 23.7567 | 36.238 | -2.059 | 2.1934 |
| $3 \times 10^{3}$ | 0.7957 | 0.5115 | -0.4798 | 0.9888 | 0.8181 | 0.9599 | 1.4549 | -0.0288 | 0.2297 |
| $5 \times 10^{3}$ | 1.0346 | 1.0412 | 1.0163 | 1.2809 | 0.9269 | 0.9866 | 1.2262 | 0.7647 | 0.8542 |
| $10^{4}$ | 1.0479 | 1.0848 | 1.1772 | 1.2601 | 0.9551 | 0.9977 | 1.1615 | 0.8816 | 0.9377 |


| G | $2 \times 10^{3}$ | $3 \times 10^{3}$ | $5 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| N | 1 | 1 | 1 | 2 | -.5 | -.8 | 1 | 1 | 1 |
| Sc | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 2.01 | 0.24 | 0.6 |

Table. 7
Nusselt Number (Nu) at $\mathrm{y}=1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | -32.685 | -55.004 | 19.548 | -87.091 | -21.091 | -21.162 | -12.847 | 1.047 | 18.514 |
| $3 \times 10^{3}$ | -4.058 | -6.8767 | -3.079 | -8.721 | -5.598 | -7.381 | -0.3126 | 0.0197 | 0.1105 |
| $5 \times 10^{3}$ | -3.273 | -4.147 | -2.8799 | -4.627 | -4.412 | -5.641 | 0.3753 | 1.0609 | 1.656 |
| $10^{4}$ | -3.165 | -3.731 | -2.8912 | -4.022 | -4.246 | -5.387 | 0.5461 | 1.3211 | 2.0347 |

Table. 8
Nusselt Number (Nu) at $\mathrm{y}=-1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | 19.551 | -26.4058 | 51.4558 | -40.4566 | 0.8943 | -0.9451 | 61.1645 | 76.5728 | 86.2831 |
| $3 \times 10^{3}$ | 0.7956 | -0.8854 | 2.0681 | -2.2941 | 1.4241 | 1.9545 | -2.7001 | -3.8194 | -5.0369 |
| $5 \times 10^{3}$ | 1.0346 | 0.4073 | 1.3921 | 0.1376 | 1.7679 | 2.3462 | -3.4496 | -4.9583 | -6.6221 |
| $10^{4}$ | 1.0479 | 0.6473 | 1.2573 | 0.4561 | 1.8468 | 2.5276 | -3.3301 | -4.7203 | -6.2286 |


| $\mathrm{S}_{0}$ | -.05 | 1 | -1 | 0.5 | 0.5 | 0.5 | 0.5 | 0.5 | 0.5 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $\alpha$ | 2 | 2 | 2 | 2 | 5 | 10 | -2 | -5 | -10 |

Table. 9
Sherwood Number (Sh) at $\mathrm{y}=1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | 16.995 | 15.641 | 29.837 | 8.754 | -35.546 | -25.905 | -2.588 | 7.9616 | 16.482 |
| $3 \times 10^{3}$ | -1.612 | -.6639 | 2.569 | -1.047 | .8189 | .4303 | -2.519 | -.992 | -1.088 |
| $5 \times 10^{3}$ | -2.122 | -1.902 | -1.197 | -1.477 | 1.477 | .6035 | -2.955 | -1.166 | -1.451 |


| $10^{4}$ | -2.192 | -2.062 | -1.645 | -1.538 | 1.527 | -.9751 | -2.961 | -1.193 | -1.505 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |

Table. 10
Sherwood Number (Sh) at $\mathrm{y}=-1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | -11.758 | -2.898 | 9.589 | -12.092 | -12.973 | 10.865 | -14.401 | -.3928 | -1.551 |
| $3 \times 10^{3}$ | .4328 | .6175 | 1.262 | -.3464 | -3.836 | -2.657 | .5528 | -.6365 | -.1689 |
| $5 \times 10^{3}$ | .2775 | .2732 | .2894 | -.4413 | -3.695 | -2.636 | .7827 | -.7318 | -.3563 |
| $10^{4}$ | .2688 | .2448 | .1848 | -.4345 | -3.658 | -1.016 | .8476 | -.7458 | -.3813 |


| G | $2 \times 10^{3}$ | $3 \times 10^{3}$ | $5 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ | $2 \times 10^{3}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| N | 1 | 1 | 1 | 2 | -.5 | -.8 | 1 | 1 | 1 |
| Sc | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 2.01 | 0.24 | 0.6 |

Table. 11
Sherwood Number (Sh) at $\mathrm{y}=1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | 16.995 | -19.503 | $-12,913$ | -35.712 | 13.123 | 6.905 | 9.311 | 1.569 | -8.484 |
| $3 \times 10^{3}$ | -1.612 | -2.221 | -3.496 | -5.837 | -1.911 | -2.053 | 1.153 | 2.237 | 3.478 |
| $5 \times 10^{3}$ | -2.122 | -1.123 | -3.756 | -.5154 | -2.683 | -3.184 | .7061 | 1.561 | 2.474 |
| $10^{4}$ | -2.193 | -.1745 | -3.741 | .2717 | -2.791 | -3.348 | .5951 | 1.391 | 2.228 |

Table. 12
Sherwood Number (Sh) at $\mathrm{y}=-1$

| $\mathrm{D}^{-1}$ | I | II | III | IV | V | VI | VII | VIII | IX |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{3}$ | -11.758 | -20.116 | -19.399 | -26.494 | 0.5263 | 4.164 | -36.736 | -45.287 | -56.671 |
| $3 \times 10^{3}$ | .4328 | -3.525 | .2114 | -6.582 | 1.325 | 2.278 | -2.495 | -3.069 | -3.576 |
| $5 \times 10^{3}$ | .2775 | -2.995 | 1.091 | -4.721 | 1.102 | 2.025 | -2.007 | -2.327 | -2.546 |
| $10^{4}$ | .2688 | -2.529 | 1.265 | -4.307 | 1.049 | 1.907 | -2.085 | -2.482 | -2.802 |


| $\mathrm{S}_{0}$ | -.05 | 1 | -1 | 0.5 | 0.5 | 0.5 | 0.5 | 0.5 | 0.5 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $\alpha$ | 2 | 2 | 2 | 2 | 5 | 10 | -2 | -5 | -10 |

## IV. CONCLUSIONS

The soret effect on convective heat and mass transfer flow of viscous incompressible fluid through a porous medium in a vertical channel have been analysed.For all variations we find that the velocity gradually rises from its prescribed value at $\mathrm{y}=-1$ to attain a maximum at $\mathrm{y}=-0.6$ and then falls to its prescribed value at $\mathrm{y}=1$. The temperature and concentration is positive or negative according as the actual temperature or concentration is greater or smaller than the mean temperature and concentration respectively. We find that the shear stress on the boundaries are almost negative except for $\alpha<0$. The rate of heat transfer decreases with lower permeability and enhances with higher permeability for various parameters while Sherwood number reduces at the plate $y=1$ and enhances at $\mathrm{y}=-1$ for various parameters.

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